

(*N*-Phenyl-2-[2'-[(phenylethylcarbamoyl)methoxy]biphenyl-2-yloxy]-*N*-ethylacetamide- $\kappa^4$ O)(2,4,6-trinitrophenolato- $\kappa$ O)bis(2,4,6-trinitrophenolato- $\kappa^2$ O, $O'$ )-gadolinium(III)

Yan-Ling Guo,<sup>a</sup> Ya-Wen Wang,<sup>a</sup> Wei Dou,<sup>a</sup> Wei-Sheng Liu<sup>a\*</sup> and Da-Qi Wang<sup>b</sup>

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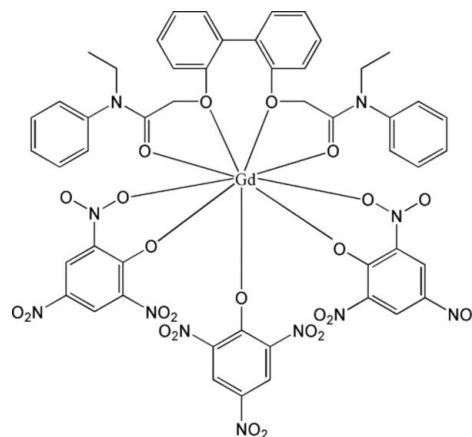
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Key indicators: single-crystal X-ray study;  $T = 298$  K; mean  $\sigma(\text{C}-\text{C}) = 0.008$  Å;  $R$  factor = 0.036;  $wR$  factor = 0.086; data-to-parameter ratio = 12.2.

In the title compound,  $[\text{Gd}(\text{C}_6\text{H}_2\text{N}_3\text{O}_7)_3(\text{C}_{32}\text{H}_{32}\text{N}_2\text{O}_4)]$ , the  $\text{Gd}^{\text{III}}$  ion is nine-coordinated by four O atoms from the *N*-phenyl-2-[2'-[(phenylethylcarbamoyl)methoxy]biphenyl-2-yloxy]-*N*-ethylacetamide ligand and five from two bidentate and one monodentate 2,4,6-trinitrophenolate (picrate) ligands. The coordination polyhedron around the  $\text{Gd}^{\text{III}}$  ion is a distorted tricapped trigonal prism. The tetradentate ligand wraps around the metal ion with its O atoms and forms a ring-like coordination structure together with the Gd atom. The mean distance between the Gd atom and the coordinated carbonyl O atoms is shorter than that between the Gd atom and the coordinated ether atoms. This suggests that the Gd—O (C=O) bond is stronger than the Gd—O (C—O—C) bond.

## Related literature

For general background, see: Lehn, 1995; Steed & Atwood, 2000; Bourne *et al.*, 2001; Evan & Lin, 2002; Janiak, 2003; Carlucci *et al.*, 2003; Gutschke *et al.*, 1996. For related structures, see: Wang *et al.*, 2003.



## Experimental

### Crystal data

$[\text{Gd}(\text{C}_6\text{H}_2\text{N}_3\text{O}_7)_3(\text{C}_{32}\text{H}_{32}\text{N}_2\text{O}_4)]$   
 $M_r = 1350.16$   
Monoclinic,  $P2_1/c$   
 $a = 20.125$  (2) Å  
 $b = 16.7778$  (18) Å  
 $c = 16.6491$  (18) Å  
 $\beta = 105.051$  (2)°

$V = 5428.9$  (10) Å<sup>3</sup>  
 $Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 1.32$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
 $0.40 \times 0.37 \times 0.29$  mm

### Data collection

Bruker SMART CCD area detector diffractometer  
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  
 $T_{\text{min}} = 0.620$ ,  $T_{\text{max}} = 0.701$

28190 measured reflections  
9559 independent reflections  
6585 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.036$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.087$   
 $S = 1.08$   
9559 reflections

786 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 1.50$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.58$  e Å<sup>-3</sup>

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

We are grateful to the NSFC (grants 20371022, 20431010 and 20021001), the Specialized Research Fund for the Doctoral Program of Higher Education (200307300015), and the Key Project of the Ministry of Education of China (grant 01170) for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DN2162).

## References

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**supplementary materials**

*Acta Cryst.* (2007). E63, m1619-m1620 [ doi:10.1107/S160053680702137X ]

**(*N*-Phenyl-2-{2'-[(phenylethylcarbamoyl)methoxy]biphenyl-2-yloxy}-*N*-ethylacetamide- $\kappa^4O$ )(2,4,6-trinitrophenolato- $\kappa O$ )bis(2,4,6-trinitrophenolato- $\kappa^2O,O'$ )gadolinium(III)**

**Y.-L. Guo, Y.-W. Wang, W. Dou, W.-S. Liu and D.-Q. Wang**

### Comment

Metal-directed self-assembly of organic ligands and metal ions, or organometallic moieties, to form well defined structures has been a very attractive field in recent years (Lehn, 1995; Steed & Atwood, 2000; Bourne *et al.*, 2001;), for their fascinating structural diversities and potential application in functional materials, nanotechnology and biological recognition (Evan & Lin, 2002; Janiak, 2003; Carlucci *et al.*, 2003; Gutschke *et al.*, 1996;). We have now investigated the assembly of the ligand *N*-phenyl-2-{2'-[(phenyl-ethyl-carbamoyl)-methoxy]-biphenyl-2-yloxy}-*N*-ethyl-acetamide with gadolinium picrate.

In the title complex I, The Gd(III) ion is coordinated by 9 oxygen donor atoms, five of them belonging to two bidentate and one monodentate picrate groups whereas the remaining four are from the tetradentate ligand *L* (Fig. 1). The coordination polyhedron around Gd(III) ion is a distorted tricapped trigonal prism (Fig. 2) (Wang *et al.*, 2003). The quadridentate ligand wraps around the metal ion with its oxygen atoms and forms a ring-like coordination structure together with the Gd atom. The two phenyl rings about the central bond in the molecule have a drastic twisting, with the dihedral angle between them being 61.18°. The mean distances between the Gd atom and the coordinated ether and carbonyl oxygen atoms are 2.575 Å and 2.3105 Å, respectively. This suggests that the Gd—O (C=O) bond is stronger than the Gd—O (C—O—C) bond.

### Experimental

To a solution of 0.2 mmol gadolinium picrate in 5 ml of ethanol was added dropwise the solution of 0.2 mmol ligand in 10 ml of ethanol. The mixture was stirred at room temperature for 6 h. The precipitated solid complex was filtered, washed with ethanol and dried *in vacuo* over P<sub>2</sub>O<sub>5</sub> for 72 h. The complex was obtained as yellow powder. Single crystal of the gadolinium complex grew from CH<sub>3</sub>Cl and CH<sub>3</sub>CH<sub>2</sub>OH by slow evaporation at room temperature. After about two weeks, yellow crystal formed from the solution.

### Refinement

All H atoms attached were fixed geometrically and treated as riding with C—H = 0.93 Å (C<sub>aromatic</sub>), 0.97 Å (C<sub>methylene</sub>) and C—H = 0.96 Å (C<sub>methyl</sub>) with  $U_{iso}(H) = 1.2U_{eq}(C_{aromatic} \text{ or } C_{methylene})$  and  $U_{iso}(H) = 1.5U_{eq}(C_{methyl})$ .

Figures

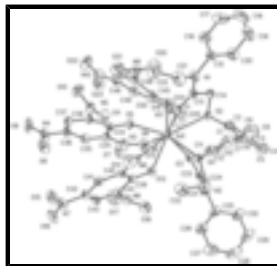


Fig. 1. The independent components of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms have been omitted for clarity.

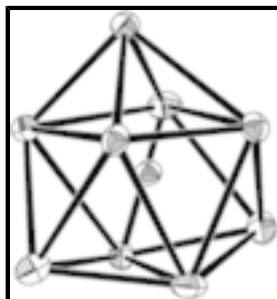


Fig. 2. The coordination polyhedron of (I), showing the distorted tricapped trigonal prism.

(*N*-Phenyl-2-{2'-[(phenylethylcarbamoyl)methoxy]biphenyl-2-yloxy}-*N*-ethylacetamide- $\kappa^4 O$ )(2,4,6-trinitrophenolato- $\kappa O$ ) bis(2,4,6-trinitrophenolato- $\kappa^2 O, O'$ )gadolinium(III)

Crystal data

[Gd(C<sub>6</sub>H<sub>2</sub>N<sub>3</sub>O<sub>7</sub>)<sub>3</sub>(C<sub>32</sub>H<sub>32</sub>N<sub>2</sub>O<sub>4</sub>)]

$M_r = 1350.16$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 20.125 (2) \text{ \AA}$

$b = 16.7778 (18) \text{ \AA}$

$c = 16.6491 (18) \text{ \AA}$

$\beta = 105.051 (2)^\circ$

$V = 5428.9 (10) \text{ \AA}^3$

$Z = 4$

$F_{000} = 2716$

$D_x = 1.652 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation

$\lambda = 0.71073 \text{ \AA}$

Cell parameters from 8109 reflections

$\theta = 2.2\text{--}25.3^\circ$

$\mu = 1.32 \text{ mm}^{-1}$

$T = 298 (2) \text{ K}$

Block, yellow

$0.40 \times 0.37 \times 0.29 \text{ mm}$

Data collection

Bruker SMART CCD area detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 298(2) \text{ K}$

phi and  $\omega$  scans

Absorption correction: multi-scan  
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.620$ ,  $T_{\max} = 0.701$

9559 independent reflections

6585 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.036$

$\theta_{\text{max}} = 25.0^\circ$

$\theta_{\text{min}} = 1.8^\circ$

$h = -16 \rightarrow 23$

$k = -19 \rightarrow 19$

28190 measured reflections

$l = -19 \rightarrow 17$

*Refinement*

Refinement on  $F^2$

Secondary atom site location: difference Fourier map

Least-squares matrix: full

Hydrogen site location: inferred from neighbouring sites

$R[F^2 > 2\sigma(F^2)] = 0.036$

H-atom parameters constrained

$wR(F^2) = 0.087$

$$w = 1/[\sigma^2(F_o^2) + (0.0225P)^2 + 8.0112P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$S = 1.08$

$(\Delta/\sigma)_{\max} = 0.002$

9559 reflections

$\Delta\rho_{\max} = 1.50 \text{ e } \text{\AA}^{-3}$

786 parameters

$\Delta\rho_{\min} = -0.58 \text{ e } \text{\AA}^{-3}$

Primary atom site location: structure-invariant direct methods

Extinction correction: none

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Gd	0.749509 (11)	0.628330 (12)	0.769967 (14)	0.03694 (8)
N1	0.9530 (2)	0.7437 (2)	0.8655 (3)	0.0527 (11)
N2	0.5485 (2)	0.7472 (3)	0.6819 (3)	0.0604 (12)
N3	0.6661 (2)	0.4428 (3)	0.7896 (3)	0.0558 (11)
N4	0.7446 (4)	0.1794 (3)	0.7376 (3)	0.0817 (17)
N5	0.8962 (2)	0.4044 (3)	0.7360 (3)	0.0580 (11)
N6	0.8047 (2)	0.5775 (2)	0.5830 (3)	0.0526 (11)
N7	0.7101 (3)	0.3392 (3)	0.4317 (3)	0.0794 (15)
N8	0.5725 (2)	0.4815 (3)	0.5851 (3)	0.0649 (13)
N9	0.9043 (3)	0.5088 (3)	0.9422 (3)	0.0657 (13)
N10	0.7864 (4)	0.3033 (3)	1.0554 (3)	0.0798 (16)
N11	0.6777 (3)	0.5639 (3)	0.9963 (4)	0.0712 (14)
O1	0.85746 (15)	0.68313 (18)	0.7941 (2)	0.0476 (8)
O2	0.77451 (14)	0.75760 (16)	0.85945 (19)	0.0421 (8)
O3	0.64424 (14)	0.69007 (17)	0.75993 (19)	0.0411 (7)
O4	0.72718 (14)	0.75474 (16)	0.68295 (18)	0.0398 (7)
O5	0.79637 (15)	0.50497 (16)	0.7700 (2)	0.0456 (8)

## supplementary materials

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O6	0.66987 (17)	0.51556 (19)	0.7911 (2)	0.0602 (10)
O7	0.6175 (2)	0.4090 (2)	0.8044 (3)	0.1057 (17)
O8	0.6861 (3)	0.1556 (2)	0.7312 (3)	0.1065 (17)
O9	0.7935 (3)	0.1368 (2)	0.7345 (3)	0.1079 (17)
O10	0.9479 (2)	0.3659 (2)	0.7687 (3)	0.0836 (12)
O11	0.8964 (2)	0.4630 (2)	0.6944 (3)	0.0776 (12)
O12	0.68209 (16)	0.58803 (18)	0.6408 (2)	0.0524 (9)
O13	0.81129 (17)	0.62156 (19)	0.6441 (2)	0.0576 (9)
O14	0.8431 (2)	0.5811 (2)	0.5371 (3)	0.0747 (11)
O15	0.7645 (3)	0.3316 (3)	0.4143 (3)	0.1159 (19)
O16	0.6607 (3)	0.2947 (3)	0.4072 (3)	0.0965 (15)
O17	0.5489 (2)	0.4223 (3)	0.6085 (3)	0.1105 (17)
O18	0.5445 (2)	0.5463 (3)	0.5762 (3)	0.0924 (14)
O19	0.78189 (17)	0.60133 (18)	0.9109 (2)	0.0557 (9)
O20	0.9221 (2)	0.5779 (3)	0.9556 (3)	0.0882 (13)
O21	0.9358 (2)	0.4588 (3)	0.9138 (3)	0.0891 (13)
O22	0.8388 (3)	0.2635 (3)	1.0675 (3)	0.1086 (17)
O23	0.7321 (3)	0.2800 (2)	1.0683 (3)	0.0938 (14)
O24	0.6489 (2)	0.5496 (3)	1.0510 (3)	0.0947 (15)
O25	0.6636 (3)	0.6196 (3)	0.9512 (4)	0.135 (2)
C1	0.7289 (2)	0.8213 (3)	0.8567 (3)	0.0438 (12)
C2	0.6850 (3)	0.8186 (3)	0.9071 (3)	0.0571 (14)
H2	0.6882	0.7775	0.9455	0.069*
C3	0.6358 (3)	0.8777 (4)	0.9003 (4)	0.0713 (17)
H3	0.6048	0.8759	0.9330	0.086*
C4	0.6330 (3)	0.9391 (4)	0.8450 (4)	0.081 (2)
H4	0.6005	0.9792	0.8413	0.097*
C5	0.6779 (3)	0.9420 (3)	0.7950 (4)	0.0713 (17)
H5	0.6759	0.9845	0.7587	0.086*
C6	0.7264 (2)	0.8814 (3)	0.7984 (3)	0.0495 (12)
C7	0.7730 (2)	0.8811 (3)	0.7422 (3)	0.0495 (12)
C8	0.8195 (3)	0.9428 (3)	0.7435 (4)	0.0711 (17)
H8	0.8203	0.9863	0.7784	0.085*
C9	0.8642 (3)	0.9402 (4)	0.6938 (5)	0.083 (2)
H9	0.8956	0.9814	0.6963	0.099*
C10	0.8633 (3)	0.8781 (3)	0.6406 (4)	0.0692 (16)
H10	0.8940	0.8769	0.6074	0.083*
C11	0.8165 (2)	0.8170 (3)	0.6364 (3)	0.0546 (13)
H11	0.8153	0.7745	0.6001	0.066*
C12	0.7720 (2)	0.8196 (3)	0.6860 (3)	0.0421 (11)
C13	0.8865 (2)	0.7291 (3)	0.8501 (3)	0.0404 (11)
C14	0.8455 (2)	0.7713 (3)	0.9012 (3)	0.0466 (12)
H14A	0.8562	0.7498	0.9572	0.056*
H14B	0.8556	0.8279	0.9043	0.056*
C15	0.9900 (2)	0.7912 (3)	0.9347 (3)	0.0534 (13)
C16	1.0082 (3)	0.7587 (4)	1.0134 (4)	0.0774 (18)
H16	0.9959	0.7067	1.0226	0.093*
C17	1.0449 (4)	0.8047 (6)	1.0778 (5)	0.112 (3)
H17	1.0572	0.7841	1.1314	0.135*

C18	1.0633 (4)	0.8798 (6)	1.0638 (6)	0.114 (3)
H18	1.0885	0.9100	1.1081	0.137*
C19	1.0456 (3)	0.9123 (4)	0.9861 (6)	0.093 (2)
H19	1.0586	0.9641	0.9774	0.112*
C20	1.0086 (3)	0.8675 (3)	0.9206 (4)	0.0669 (15)
H20	0.9962	0.8887	0.8672	0.080*
C21	0.9901 (3)	0.7119 (4)	0.8036 (5)	0.093 (2)
H21A	0.9609	0.7192	0.7478	0.111*
H21B	1.0317	0.7427	0.8083	0.111*
C22	1.0076 (4)	0.6313 (4)	0.8152 (5)	0.115 (3)
H22A	1.0355	0.6233	0.8709	0.173*
H22B	1.0329	0.6155	0.7763	0.173*
H22C	0.9665	0.5999	0.8065	0.173*
C23	0.6152 (2)	0.7319 (3)	0.6994 (3)	0.0406 (11)
C24	0.6557 (2)	0.7696 (3)	0.6443 (3)	0.0429 (11)
H24A	0.6426	0.7462	0.5892	0.052*
H24B	0.6469	0.8265	0.6394	0.052*
C25	0.5145 (2)	0.7904 (3)	0.6082 (4)	0.0600 (15)
C26	0.4971 (3)	0.7523 (4)	0.5331 (4)	0.0774 (18)
H26	0.5078	0.6988	0.5295	0.093*
C27	0.4633 (3)	0.7942 (5)	0.4619 (5)	0.097 (2)
H27	0.4509	0.7688	0.4105	0.117*
C28	0.4486 (3)	0.8726 (6)	0.4686 (5)	0.102 (3)
H28	0.4264	0.9009	0.4212	0.122*
C29	0.4658 (3)	0.9104 (4)	0.5432 (5)	0.096 (2)
H29	0.4549	0.9639	0.5466	0.115*
C30	0.4995 (3)	0.8692 (4)	0.6143 (4)	0.0760 (17)
H30	0.5118	0.8950	0.6655	0.091*
C31	0.5076 (3)	0.7199 (4)	0.7423 (4)	0.0829 (19)
H31A	0.5353	0.7265	0.7991	0.100*
H31B	0.4666	0.7525	0.7351	0.100*
C32	0.4882 (4)	0.6369 (5)	0.7274 (5)	0.113 (3)
H32A	0.4636	0.6299	0.6702	0.169*
H32B	0.4592	0.6215	0.7624	0.169*
H32C	0.5288	0.6043	0.7398	0.169*
C33	0.7823 (2)	0.4308 (3)	0.7624 (3)	0.0403 (11)
C34	0.7199 (2)	0.3955 (3)	0.7704 (3)	0.0435 (11)
C35	0.7077 (3)	0.3137 (3)	0.7636 (3)	0.0553 (13)
H35	0.6665	0.2925	0.7694	0.066*
C36	0.7573 (3)	0.2654 (3)	0.7482 (3)	0.0585 (15)
C37	0.8195 (3)	0.2949 (3)	0.7425 (3)	0.0569 (14)
H37	0.8536	0.2605	0.7351	0.068*
C38	0.8307 (2)	0.3745 (3)	0.7477 (3)	0.0446 (11)
C39	0.6896 (2)	0.5324 (3)	0.5928 (3)	0.0420 (11)
C40	0.7487 (2)	0.5201 (3)	0.5628 (3)	0.0422 (11)
C41	0.7559 (3)	0.4577 (3)	0.5115 (3)	0.0529 (13)
H41	0.7957	0.4524	0.4936	0.063*
C42	0.7036 (3)	0.4038 (3)	0.4877 (3)	0.0548 (14)
C43	0.6442 (3)	0.4111 (3)	0.5133 (3)	0.0537 (13)



## supplementary materials

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H43	0.6092	0.3736	0.4974	0.064*
C44	0.6373 (2)	0.4742 (3)	0.5623 (3)	0.0463 (12)
C45	0.7854 (3)	0.5374 (3)	0.9522 (3)	0.0495 (13)
C46	0.8420 (3)	0.4827 (3)	0.9640 (3)	0.0503 (13)
C47	0.8420 (3)	0.4077 (3)	0.9951 (3)	0.0576 (14)
H47	0.8786	0.3733	0.9974	0.069*
C48	0.7870 (3)	0.3842 (3)	1.0228 (3)	0.0586 (14)
C49	0.7337 (3)	0.4356 (3)	1.0239 (3)	0.0597 (14)
H49	0.6985	0.4200	1.0472	0.072*
C50	0.7336 (3)	0.5102 (3)	0.9904 (3)	0.0539 (13)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Gd	0.03632 (13)	0.02908 (11)	0.04381 (13)	0.00121 (11)	0.00754 (9)	-0.00233 (11)
N1	0.037 (2)	0.055 (3)	0.070 (3)	-0.009 (2)	0.021 (2)	-0.017 (2)
N2	0.037 (2)	0.070 (3)	0.076 (3)	0.010 (2)	0.017 (2)	0.027 (2)
N3	0.060 (3)	0.051 (3)	0.061 (3)	-0.011 (2)	0.024 (2)	0.003 (2)
N4	0.141 (5)	0.024 (2)	0.083 (4)	-0.008 (3)	0.034 (4)	0.005 (2)
N5	0.058 (3)	0.053 (3)	0.068 (3)	0.014 (2)	0.025 (2)	-0.007 (2)
N6	0.055 (3)	0.052 (3)	0.049 (3)	0.006 (2)	0.010 (2)	0.010 (2)
N7	0.082 (4)	0.082 (4)	0.067 (4)	0.012 (3)	0.006 (3)	-0.030 (3)
N8	0.056 (3)	0.078 (4)	0.059 (3)	-0.008 (3)	0.013 (2)	-0.019 (3)
N9	0.068 (3)	0.074 (3)	0.053 (3)	0.006 (3)	0.013 (2)	0.007 (3)
N10	0.110 (5)	0.067 (4)	0.062 (3)	-0.006 (4)	0.021 (3)	0.016 (3)
N11	0.075 (4)	0.057 (3)	0.085 (4)	-0.009 (3)	0.028 (3)	-0.014 (3)
O1	0.0425 (19)	0.0457 (18)	0.055 (2)	-0.0046 (15)	0.0132 (15)	-0.0132 (16)
O2	0.0371 (18)	0.0340 (16)	0.053 (2)	0.0047 (13)	0.0076 (14)	-0.0029 (14)
O3	0.0390 (18)	0.0380 (17)	0.047 (2)	0.0055 (14)	0.0124 (15)	0.0019 (15)
O4	0.0328 (17)	0.0326 (16)	0.053 (2)	0.0002 (13)	0.0082 (14)	0.0022 (14)
O5	0.0463 (19)	0.0267 (16)	0.067 (2)	0.0026 (14)	0.0214 (16)	-0.0031 (15)
O6	0.057 (2)	0.0381 (19)	0.093 (3)	0.0003 (17)	0.032 (2)	0.0017 (18)
O7	0.090 (3)	0.070 (3)	0.186 (5)	-0.021 (2)	0.087 (3)	-0.001 (3)
O8	0.148 (5)	0.054 (3)	0.123 (4)	-0.038 (3)	0.044 (4)	-0.008 (2)
O9	0.167 (5)	0.037 (2)	0.123 (4)	0.029 (3)	0.043 (3)	-0.003 (2)
O10	0.062 (3)	0.086 (3)	0.103 (3)	0.030 (2)	0.022 (2)	0.005 (3)
O11	0.075 (3)	0.064 (3)	0.109 (3)	0.010 (2)	0.052 (2)	0.017 (2)
O12	0.056 (2)	0.0490 (19)	0.048 (2)	0.0068 (16)	0.0069 (16)	-0.0169 (16)
O13	0.061 (2)	0.048 (2)	0.060 (2)	-0.0081 (18)	0.0088 (17)	-0.0074 (19)
O14	0.077 (3)	0.084 (3)	0.070 (3)	-0.011 (2)	0.032 (2)	0.004 (2)
O15	0.089 (4)	0.146 (5)	0.108 (4)	0.025 (3)	0.018 (3)	-0.072 (3)
O16	0.111 (4)	0.080 (3)	0.091 (3)	-0.011 (3)	0.013 (3)	-0.047 (3)
O17	0.102 (4)	0.116 (4)	0.132 (5)	-0.034 (3)	0.064 (3)	-0.001 (3)
O18	0.058 (3)	0.103 (4)	0.114 (4)	0.010 (3)	0.019 (2)	-0.039 (3)
O19	0.076 (2)	0.0409 (19)	0.048 (2)	0.0006 (17)	0.0120 (17)	0.0030 (16)
O20	0.084 (3)	0.082 (3)	0.100 (4)	-0.020 (3)	0.025 (2)	0.006 (3)
O21	0.077 (3)	0.104 (3)	0.092 (3)	0.019 (3)	0.033 (3)	0.006 (3)
O22	0.130 (4)	0.076 (3)	0.121 (4)	0.031 (3)	0.034 (3)	0.041 (3)

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O23	0.128 (4)	0.072 (3)	0.082 (3)	-0.026 (3)	0.027 (3)	0.015 (2)
O24	0.093 (3)	0.084 (3)	0.124 (4)	-0.014 (3)	0.059 (3)	-0.024 (3)
O25	0.138 (5)	0.114 (4)	0.176 (6)	0.071 (4)	0.081 (4)	0.067 (4)
C1	0.038 (3)	0.033 (3)	0.056 (3)	0.003 (2)	0.005 (2)	-0.016 (2)
C2	0.052 (3)	0.060 (3)	0.060 (4)	0.000 (3)	0.016 (3)	-0.021 (3)
C3	0.056 (4)	0.082 (4)	0.077 (4)	0.004 (3)	0.018 (3)	-0.035 (4)
C4	0.064 (4)	0.077 (4)	0.094 (5)	0.027 (3)	0.007 (4)	-0.038 (4)
C5	0.075 (4)	0.045 (3)	0.085 (5)	0.016 (3)	0.005 (4)	-0.008 (3)
C6	0.048 (3)	0.029 (3)	0.066 (3)	0.005 (2)	0.005 (3)	-0.007 (2)
C7	0.050 (3)	0.031 (3)	0.062 (3)	-0.002 (2)	0.005 (3)	0.006 (2)
C8	0.086 (5)	0.036 (3)	0.086 (5)	-0.016 (3)	0.013 (4)	0.006 (3)
C9	0.076 (5)	0.059 (4)	0.106 (6)	-0.034 (3)	0.010 (4)	0.024 (4)
C10	0.056 (3)	0.067 (4)	0.087 (5)	-0.011 (3)	0.023 (3)	0.027 (4)
C11	0.053 (3)	0.054 (3)	0.058 (4)	-0.001 (3)	0.015 (3)	0.013 (3)
C12	0.035 (3)	0.035 (3)	0.053 (3)	-0.002 (2)	0.005 (2)	0.010 (2)
C13	0.039 (3)	0.035 (2)	0.048 (3)	-0.003 (2)	0.014 (2)	-0.002 (2)
C14	0.039 (3)	0.049 (3)	0.050 (3)	0.000 (2)	0.007 (2)	-0.011 (2)
C15	0.034 (3)	0.058 (3)	0.066 (4)	-0.009 (2)	0.011 (2)	-0.013 (3)
C16	0.064 (4)	0.081 (4)	0.080 (5)	-0.008 (3)	0.007 (3)	0.000 (4)
C17	0.094 (6)	0.153 (8)	0.070 (5)	-0.008 (6)	-0.012 (4)	-0.007 (5)
C18	0.081 (5)	0.134 (8)	0.113 (7)	-0.037 (5)	0.001 (5)	-0.053 (6)
C19	0.069 (5)	0.077 (5)	0.137 (7)	-0.028 (4)	0.030 (5)	-0.030 (5)
C20	0.048 (3)	0.067 (4)	0.088 (4)	-0.012 (3)	0.023 (3)	-0.009 (3)
C21	0.057 (4)	0.085 (5)	0.147 (7)	-0.010 (4)	0.044 (4)	-0.019 (5)
C22	0.084 (5)	0.109 (6)	0.154 (8)	0.010 (5)	0.033 (5)	-0.011 (6)
C23	0.038 (3)	0.034 (2)	0.049 (3)	0.000 (2)	0.012 (2)	-0.005 (2)
C24	0.033 (3)	0.046 (3)	0.048 (3)	0.005 (2)	0.009 (2)	0.003 (2)
C25	0.034 (3)	0.068 (4)	0.078 (4)	0.008 (3)	0.015 (3)	0.019 (3)
C26	0.057 (4)	0.084 (4)	0.087 (5)	-0.003 (3)	0.010 (3)	0.012 (4)
C27	0.070 (5)	0.128 (7)	0.088 (6)	0.001 (5)	0.010 (4)	0.004 (5)
C28	0.068 (4)	0.144 (8)	0.094 (6)	0.027 (5)	0.020 (4)	0.052 (6)
C29	0.074 (5)	0.095 (5)	0.123 (7)	0.034 (4)	0.031 (5)	0.046 (5)
C30	0.054 (3)	0.082 (4)	0.094 (5)	0.021 (3)	0.021 (3)	0.021 (4)
C31	0.054 (4)	0.094 (5)	0.101 (5)	0.006 (3)	0.020 (3)	0.028 (4)
C32	0.092 (5)	0.128 (7)	0.120 (7)	-0.025 (5)	0.033 (5)	-0.014 (5)
C33	0.049 (3)	0.034 (3)	0.036 (3)	0.003 (2)	0.008 (2)	0.003 (2)
C34	0.051 (3)	0.040 (3)	0.042 (3)	0.000 (2)	0.016 (2)	0.000 (2)
C35	0.073 (4)	0.047 (3)	0.048 (3)	-0.007 (3)	0.021 (3)	0.005 (2)
C36	0.093 (4)	0.029 (2)	0.054 (4)	0.000 (3)	0.021 (3)	0.001 (2)
C37	0.075 (4)	0.044 (3)	0.051 (3)	0.016 (3)	0.016 (3)	-0.002 (2)
C38	0.056 (3)	0.032 (2)	0.047 (3)	0.005 (2)	0.015 (2)	0.000 (2)
C39	0.051 (3)	0.038 (3)	0.034 (3)	0.005 (2)	0.005 (2)	0.001 (2)
C40	0.048 (3)	0.037 (3)	0.036 (3)	0.000 (2)	0.002 (2)	0.001 (2)
C41	0.055 (3)	0.059 (3)	0.043 (3)	0.010 (3)	0.009 (2)	0.000 (3)
C42	0.067 (4)	0.054 (3)	0.040 (3)	0.007 (3)	0.008 (3)	-0.016 (2)
C43	0.067 (4)	0.042 (3)	0.048 (3)	-0.009 (3)	0.008 (3)	-0.006 (2)
C44	0.053 (3)	0.045 (3)	0.040 (3)	0.002 (2)	0.011 (2)	-0.002 (2)
C45	0.061 (3)	0.045 (3)	0.038 (3)	-0.002 (3)	0.005 (2)	-0.003 (2)
C46	0.056 (3)	0.050 (3)	0.045 (3)	-0.001 (3)	0.012 (2)	0.005 (2)

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C47	0.069 (4)	0.055 (3)	0.047 (3)	0.010 (3)	0.011 (3)	0.010 (3)
C48	0.075 (4)	0.049 (3)	0.049 (3)	0.003 (3)	0.011 (3)	0.015 (3)
C49	0.063 (4)	0.064 (4)	0.052 (3)	-0.008 (3)	0.016 (3)	0.004 (3)
C50	0.060 (3)	0.048 (3)	0.052 (3)	0.000 (3)	0.010 (3)	-0.003 (3)

### *Geometric parameters (Å, °)*

Gd—O5	2.274 (3)	C10—C11	1.380 (7)
Gd—O1	2.296 (3)	C10—H10	0.9300
Gd—O19	2.311 (3)	C11—C12	1.369 (7)
Gd—O3	2.325 (3)	C11—H11	0.9300
Gd—O12	2.326 (3)	C13—C14	1.508 (6)
Gd—O4	2.542 (3)	C14—H14A	0.9700
Gd—O6	2.562 (3)	C14—H14B	0.9700
Gd—O2	2.605 (3)	C15—C20	1.370 (7)
Gd—O13	2.703 (4)	C15—C16	1.378 (7)
N1—C13	1.318 (5)	C16—C17	1.371 (9)
N1—C15	1.440 (6)	C16—H16	0.9300
N1—C21	1.517 (7)	C17—C18	1.350 (10)
N2—C23	1.321 (5)	C17—H17	0.9300
N2—C25	1.436 (6)	C18—C19	1.363 (10)
N2—C31	1.527 (7)	C18—H18	0.9300
N3—O7	1.209 (5)	C19—C20	1.373 (8)
N3—O6	1.223 (5)	C19—H19	0.9300
N3—C34	1.445 (6)	C20—H20	0.9300
N4—O8	1.221 (6)	C21—C22	1.397 (8)
N4—O9	1.228 (7)	C21—H21A	0.9700
N4—C36	1.468 (6)	C21—H21B	0.9700
N5—O11	1.203 (5)	C22—H22A	0.9600
N5—O10	1.225 (5)	C22—H22B	0.9600
N5—C38	1.469 (6)	C22—H22C	0.9600
N6—O14	1.221 (5)	C23—C24	1.515 (6)
N6—O13	1.236 (5)	C24—H24A	0.9700
N6—C40	1.454 (6)	C24—H24B	0.9700
N7—O15	1.210 (6)	C25—C26	1.365 (8)
N7—O16	1.224 (6)	C25—C30	1.367 (7)
N7—C42	1.458 (6)	C26—C27	1.394 (9)
N8—O17	1.208 (6)	C26—H26	0.9300
N8—O18	1.216 (6)	C27—C28	1.360 (10)
N8—C44	1.453 (6)	C27—H27	0.9300
N9—O20	1.218 (6)	C28—C29	1.357 (10)
N9—O21	1.219 (6)	C28—H28	0.9300
N9—C46	1.460 (7)	C29—C30	1.386 (8)
N10—O22	1.221 (6)	C29—H29	0.9300
N10—O23	1.230 (6)	C30—H30	0.9300
N10—C48	1.462 (7)	C31—C32	1.451 (8)
N11—O25	1.186 (6)	C31—H31A	0.9700
N11—O24	1.223 (6)	C31—H31B	0.9700
N11—C50	1.466 (7)	C32—H32A	0.9600

O1—C13	1.234 (5)	C32—H32B	0.9600
O2—C1	1.402 (5)	C32—H32C	0.9600
O2—C14	1.436 (5)	C33—C38	1.424 (6)
O3—C23	1.243 (5)	C33—C34	1.426 (6)
O4—C12	1.405 (5)	C34—C35	1.393 (6)
O4—C24	1.437 (5)	C35—C36	1.360 (7)
O5—C33	1.275 (5)	C35—H35	0.9300
O12—C39	1.262 (5)	C36—C37	1.372 (7)
O19—C45	1.266 (5)	C37—C38	1.354 (6)
C1—C2	1.368 (7)	C37—H37	0.9300
C1—C6	1.391 (7)	C39—C40	1.420 (6)
C2—C3	1.385 (7)	C39—C44	1.429 (6)
C2—H2	0.9300	C40—C41	1.382 (6)
C3—C4	1.371 (8)	C41—C42	1.365 (7)
C3—H3	0.9300	C41—H41	0.9300
C4—C5	1.379 (8)	C42—C43	1.375 (7)
C4—H4	0.9300	C43—C44	1.367 (6)
C5—C6	1.400 (7)	C43—H43	0.9300
C5—H5	0.9300	C45—C50	1.429 (7)
C6—C7	1.487 (7)	C45—C46	1.436 (7)
C7—C12	1.390 (6)	C46—C47	1.361 (6)
C7—C8	1.392 (7)	C47—C48	1.363 (7)
C8—C9	1.373 (8)	C47—H47	0.9300
C8—H8	0.9300	C48—C49	1.380 (7)
C9—C10	1.364 (8)	C49—C50	1.369 (7)
C9—H9	0.9300	C49—H49	0.9300
O5—Gd—O1	89.48 (10)	C13—C14—H14B	110.6
O5—Gd—O19	79.26 (11)	H14A—C14—H14B	108.7
O1—Gd—O19	84.06 (12)	C20—C15—C16	121.0 (5)
O5—Gd—O3	140.85 (10)	C20—C15—N1	119.2 (5)
O1—Gd—O3	129.59 (10)	C16—C15—N1	119.7 (5)
O19—Gd—O3	99.89 (11)	C17—C16—C15	118.4 (6)
O5—Gd—O12	82.99 (11)	C17—C16—H16	120.8
O1—Gd—O12	125.18 (12)	C15—C16—H16	120.8
O19—Gd—O12	145.59 (12)	C18—C17—C16	120.5 (8)
O3—Gd—O12	76.19 (11)	C18—C17—H17	119.8
O5—Gd—O4	140.48 (10)	C16—C17—H17	119.8
O1—Gd—O4	77.57 (10)	C17—C18—C19	121.4 (7)
O19—Gd—O4	134.73 (10)	C17—C18—H18	119.3
O3—Gd—O4	64.17 (10)	C19—C18—H18	119.3
O12—Gd—O4	74.99 (10)	C18—C19—C20	119.2 (7)
O5—Gd—O6	66.55 (11)	C18—C19—H19	120.4
O1—Gd—O6	150.32 (11)	C20—C19—H19	120.4
O19—Gd—O6	74.79 (12)	C15—C20—C19	119.5 (6)
O3—Gd—O6	75.41 (10)	C15—C20—H20	120.3
O12—Gd—O6	71.14 (12)	C19—C20—H20	120.3
O4—Gd—O6	132.07 (10)	C22—C21—N1	113.4 (6)
O5—Gd—O2	137.64 (10)	C22—C21—H21A	108.9
O1—Gd—O2	62.18 (10)	N1—C21—H21A	108.9

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O19—Gd—O2	67.72 (10)	C22—C21—H21B	108.9
O3—Gd—O2	72.98 (9)	N1—C21—H21B	108.9
O12—Gd—O2	138.64 (10)	H21A—C21—H21B	107.7
O4—Gd—O2	67.09 (9)	C21—C22—H22A	109.5
O6—Gd—O2	124.83 (10)	C21—C22—H22B	109.5
O5—Gd—O13	71.81 (11)	H22A—C22—H22B	109.5
O1—Gd—O13	63.09 (10)	C21—C22—H22C	109.5
O19—Gd—O13	135.55 (11)	H22A—C22—H22C	109.5
O3—Gd—O13	123.85 (10)	H22B—C22—H22C	109.5
O12—Gd—O13	62.99 (11)	O3—C23—N2	121.7 (4)
O4—Gd—O13	69.02 (10)	O3—C23—C24	120.8 (4)
O6—Gd—O13	120.36 (10)	N2—C23—C24	117.4 (4)
O2—Gd—O13	114.77 (10)	O4—C24—C23	106.9 (3)
C13—N1—C15	123.1 (4)	O4—C24—H24A	110.3
C13—N1—C21	117.1 (4)	C23—C24—H24A	110.3
C15—N1—C21	119.7 (4)	O4—C24—H24B	110.3
C23—N2—C25	121.2 (4)	C23—C24—H24B	110.3
C23—N2—C31	119.1 (4)	H24A—C24—H24B	108.6
C25—N2—C31	119.7 (4)	C26—C25—C30	120.8 (6)
O7—N3—O6	120.9 (5)	C26—C25—N2	119.9 (5)
O7—N3—C34	118.6 (4)	C30—C25—N2	119.3 (6)
O6—N3—C34	120.5 (4)	C25—C26—C27	119.7 (7)
O8—N4—O9	124.7 (5)	C25—C26—H26	120.1
O8—N4—C36	117.8 (6)	C27—C26—H26	120.1
O9—N4—C36	117.5 (6)	C28—C27—C26	119.1 (7)
O11—N5—O10	123.9 (5)	C28—C27—H27	120.5
O11—N5—C38	119.4 (4)	C26—C27—H27	120.5
O10—N5—C38	116.7 (5)	C29—C28—C27	121.2 (7)
O14—N6—O13	122.4 (5)	C29—C28—H28	119.4
O14—N6—C40	117.8 (5)	C27—C28—H28	119.4
O13—N6—C40	119.8 (4)	C28—C29—C30	120.1 (7)
O15—N7—O16	124.3 (5)	C28—C29—H29	120.0
O15—N7—C42	117.8 (5)	C30—C29—H29	120.0
O16—N7—C42	117.9 (6)	C25—C30—C29	119.2 (7)
O17—N8—O18	124.4 (6)	C25—C30—H30	120.4
O17—N8—C44	118.1 (5)	C29—C30—H30	120.4
O18—N8—C44	117.5 (5)	C32—C31—N2	110.0 (6)
O20—N9—O21	124.6 (6)	C32—C31—H31A	109.7
O20—N9—C46	118.1 (5)	N2—C31—H31A	109.7
O21—N9—C46	117.3 (5)	C32—C31—H31B	109.7
O22—N10—O23	124.4 (6)	N2—C31—H31B	109.7
O22—N10—C48	118.6 (6)	H31A—C31—H31B	108.2
O23—N10—C48	117.0 (6)	C31—C32—H32A	109.5
O25—N11—O24	123.0 (6)	C31—C32—H32B	109.5
O25—N11—C50	120.3 (6)	H32A—C32—H32B	109.5
O24—N11—C50	116.7 (5)	C31—C32—H32C	109.5
C13—O1—Gd	128.6 (3)	H32A—C32—H32C	109.5
C1—O2—C14	117.0 (3)	H32B—C32—H32C	109.5
C1—O2—Gd	126.1 (2)	O5—C33—C38	121.3 (4)

C14—O2—Gd	115.5 (2)	O5—C33—C34	125.1 (4)
C23—O3—Gd	122.6 (3)	C38—C33—C34	113.5 (4)
C12—O4—C24	116.1 (3)	C35—C34—C33	122.9 (4)
C12—O4—Gd	127.5 (2)	C35—C34—N3	115.7 (4)
C24—O4—Gd	114.1 (2)	C33—C34—N3	121.4 (4)
C33—O5—Gd	143.8 (3)	C36—C35—C34	118.6 (5)
N3—O6—Gd	140.6 (3)	C36—C35—H35	120.7
C39—O12—Gd	131.5 (3)	C34—C35—H35	120.7
N6—O13—Gd	133.7 (3)	C35—C36—C37	121.7 (5)
C45—O19—Gd	132.8 (3)	C35—C36—N4	119.6 (6)
C2—C1—C6	122.7 (4)	C37—C36—N4	118.7 (5)
C2—C1—O2	118.4 (4)	C38—C37—C36	119.5 (5)
C6—C1—O2	118.7 (4)	C38—C37—H37	120.2
C1—C2—C3	119.1 (6)	C36—C37—H37	120.2
C1—C2—H2	120.5	C37—C38—C33	123.7 (5)
C3—C2—H2	120.5	C37—C38—N5	118.0 (4)
C4—C3—C2	119.8 (6)	C33—C38—N5	118.3 (4)
C4—C3—H3	120.1	O12—C39—C40	125.4 (4)
C2—C3—H3	120.1	O12—C39—C44	122.1 (4)
C3—C4—C5	120.8 (6)	C40—C39—C44	112.5 (4)
C3—C4—H4	119.6	C41—C40—C39	124.0 (4)
C5—C4—H4	119.6	C41—C40—N6	116.5 (5)
C4—C5—C6	120.6 (6)	C39—C40—N6	119.5 (4)
C4—C5—H5	119.7	C42—C41—C40	119.1 (5)
C6—C5—H5	119.7	C42—C41—H41	120.5
C1—C6—C5	116.9 (5)	C40—C41—H41	120.5
C1—C6—C7	121.5 (4)	C41—C42—C43	121.1 (5)
C5—C6—C7	121.6 (5)	C41—C42—N7	119.3 (5)
C12—C7—C8	117.1 (5)	C43—C42—N7	119.5 (5)
C12—C7—C6	121.6 (4)	C44—C43—C42	119.1 (5)
C8—C7—C6	121.3 (5)	C44—C43—H43	120.5
C9—C8—C7	120.7 (6)	C42—C43—H43	120.5
C9—C8—H8	119.7	C43—C44—C39	124.1 (5)
C7—C8—H8	119.7	C43—C44—N8	117.3 (5)
C10—C9—C8	121.0 (5)	C39—C44—N8	118.6 (4)
C10—C9—H9	119.5	O19—C45—C50	124.7 (5)
C8—C9—H9	119.5	O19—C45—C46	123.1 (5)
C9—C10—C11	119.7 (6)	C50—C45—C46	112.2 (4)
C9—C10—H10	120.2	C47—C46—C45	124.4 (5)
C11—C10—H10	120.2	C47—C46—N9	117.2 (5)
C12—C11—C10	119.4 (5)	C45—C46—N9	118.4 (4)
C12—C11—H11	120.3	C46—C47—C48	118.5 (5)
C10—C11—H11	120.3	C46—C47—H47	120.8
C11—C12—C7	122.1 (4)	C48—C47—H47	120.8
C11—C12—O4	117.9 (4)	C47—C48—C49	121.6 (5)
C7—C12—O4	119.8 (4)	C47—C48—N10	118.7 (5)
O1—C13—N1	121.4 (4)	C49—C48—N10	119.6 (6)
O1—C13—C14	119.9 (4)	C50—C49—C48	119.0 (5)
N1—C13—C14	118.7 (4)	C50—C49—H49	120.5

## supplementary materials

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O2—C14—C13	105.9 (3)	C48—C49—H49	120.5
O2—C14—H14A	110.6	C49—C50—C45	123.3 (5)
C13—C14—H14A	110.6	C49—C50—N11	117.0 (5)
O2—C14—H14B	110.6	C45—C50—N11	119.6 (5)

Fig. 1

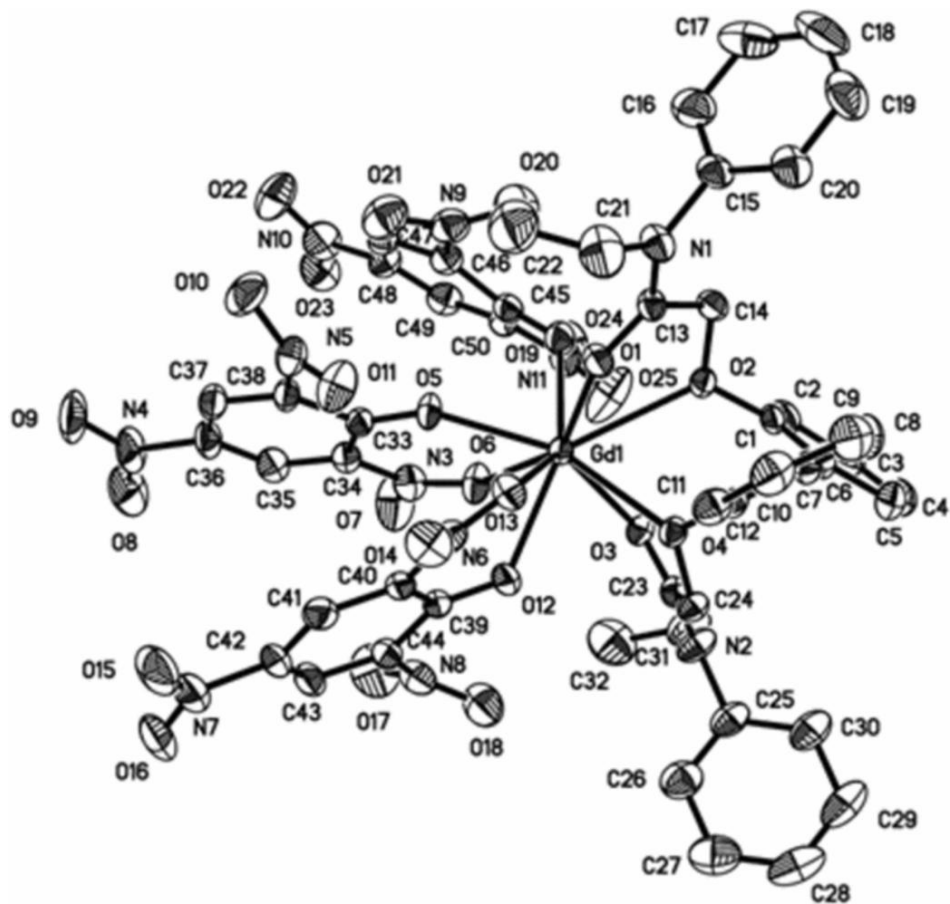




Fig. 2

